

# Electrons in solids

We have focussed so far on lattice

- lattice structure
- description of x ray scattering
- lattice vibrations

Turn now to electrons

An amazing thing is that a lot of properties of  $e^-$  in solid can be explained by ignoring lattice!

How is it possible electrons don't scatter off ions?!



Recall QM

$$\psi(\vec{r}, 0) \longrightarrow \psi(\vec{r}, t) \quad \text{How?}$$

Solve

$$\left[ -\frac{\hbar^2}{2m} \nabla^2 + V(r) \right] \phi_n(\vec{r}) = E_n \phi_n(\vec{r})$$

↑ Hermitian

↑ complete set  
↓ orthogonal

Expand

$$\psi(\vec{r}, 0) = \sum_n c_n \phi_n(\vec{r})$$

$$c_n = \int d^3\vec{r} \psi(\vec{r}, 0) \phi_n^*(\vec{r})$$

2.

$$\psi(\vec{r}, t) = \sum_n c_n \phi_n(\vec{r}) e^{-iE_n t/\hbar}$$

Sometimes eliminate  $c_n$  and write in fancy way

$$\psi(\vec{r}, t) = \int d^3 r' \left[ \sum_n \phi_n^*(r') \phi_n(r) e^{-iE_n t/\hbar} \right] \psi(\vec{r}', 0)$$



$G(\vec{r}, \vec{r}', t)$  "Green's function"  
or "propagator"

But point to emphasize is that if you have  $e^-$

in one of eigenstates initially  $\psi(\vec{r}, 0) \equiv \phi_e(\vec{r})$

if never scatters out of it  $\psi(\vec{r}, t) = \phi_e(r) e^{-iE_e t/\hbar}$

[ Analogous in a way to classical normal modes ]

One has found special states (by "absorbing"  
effect of ions into appropriate  $\phi_n(\vec{r})$ ) ~~where~~

"LANDAU FERMI LIQUID THEORY"  
 $e^-$  in solid can be understood as free electrons  
(with perhaps renormalized  $m \rightarrow m^*$ )

Perhaps  
most famous  
principle of  
CM physics

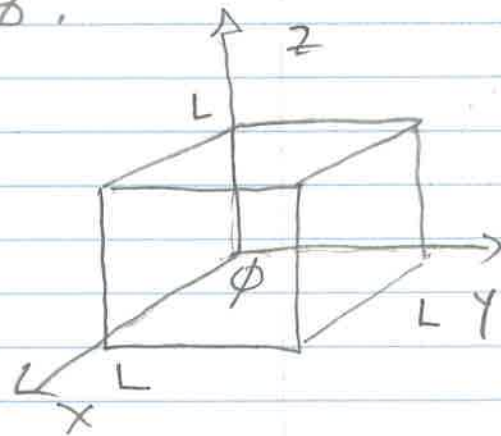
3,

We will see other ways to understand this better in the coming weeks.

For now let's see what comes out of a description of  $e^-$  in a solid as just particles in a box with no ionic potential  $V(\vec{r}) = \phi$ .

$$-\frac{\hbar^2}{2m} \nabla^2 \phi_n = E_n \phi_n$$

$$\frac{d^2}{dx^2} + \frac{d^2}{dy^2} + \frac{d^2}{dz^2}$$



Solns are

$$\phi_n(x, y, z) = \sin \frac{n_x \pi x}{L} \sin \frac{n_y \pi y}{L} \sin \frac{n_z \pi z}{L}$$

$\uparrow$   
 $n_x n_y n_z$

Can easily check this obeys Sch Eqn with

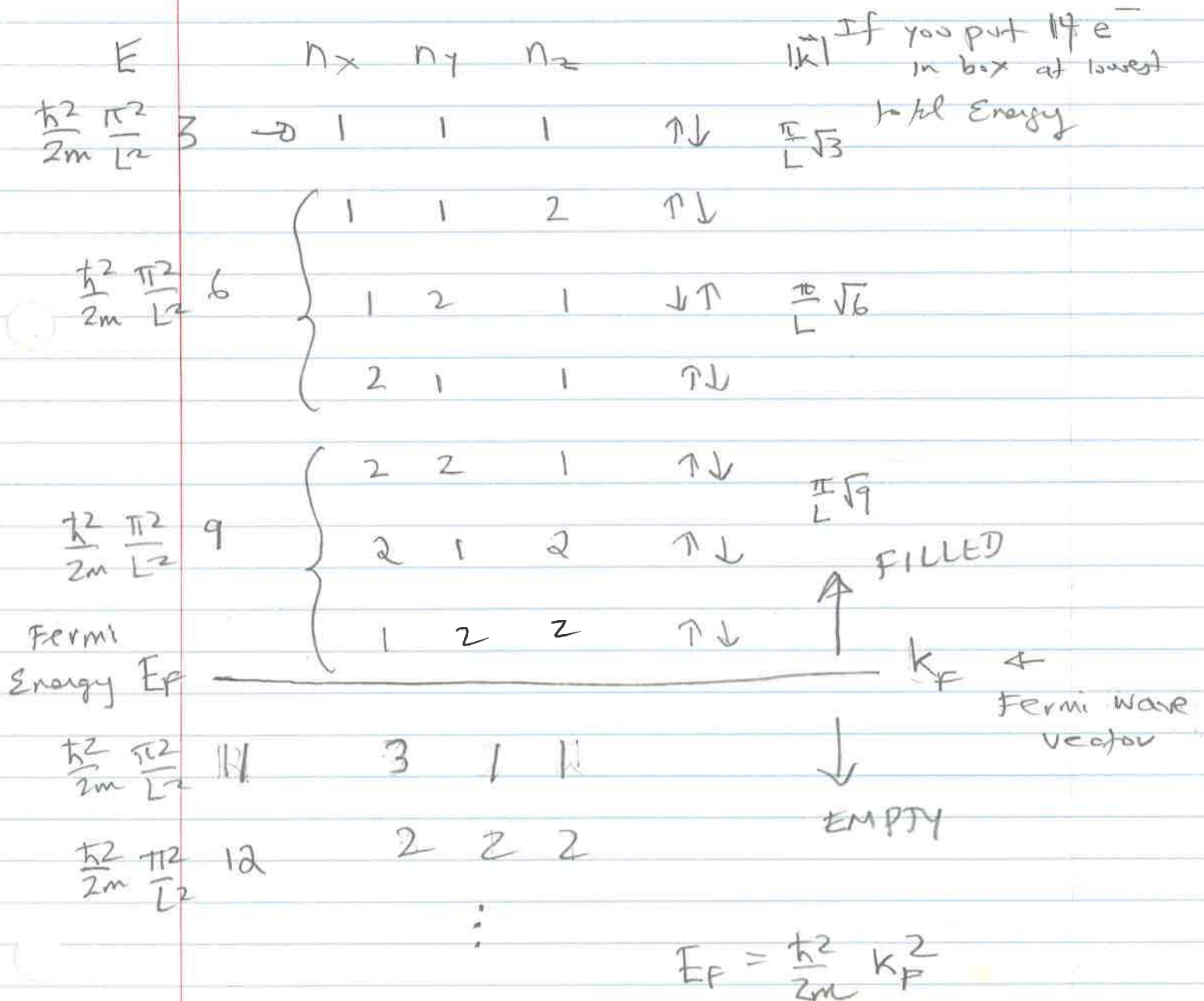
$$E_n = \frac{\hbar^2}{2m} \frac{\pi^2}{L^2} (n_x^2 + n_y^2 + n_z^2)$$

$$k_x = \frac{\pi n_x}{L} \quad k_y = \frac{\pi n_y}{L} \quad k_z = \frac{\pi n_z}{L}$$

As  $L \rightarrow \infty$   $k_x, k_y, k_z$  continuous.

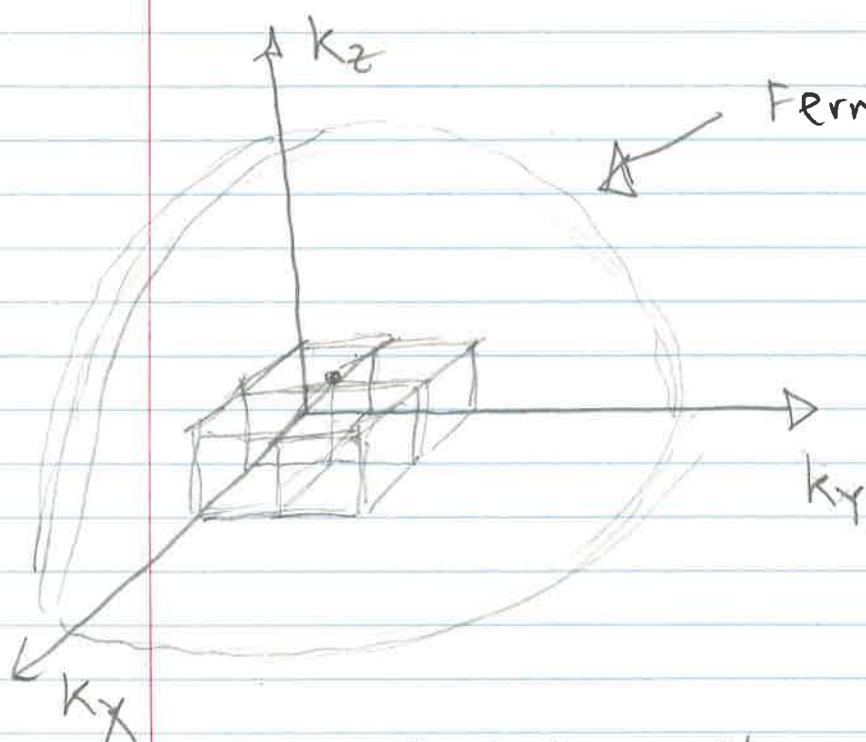
4.

Pauli Principle: For Fermions each QM state can be occupied by at most 1 particle. (or 2 if one includes spin  $\uparrow \downarrow$  as additional quantum number like  $n_x n_y n_z$ )



5.

There is a  $\vec{k}$  point in every  $\left(\frac{2\pi}{L}\right)^3$  of  $\vec{k}$  space



# points inside

$$= \frac{\frac{4}{3}\pi k_F^3}{\left(\frac{2\pi}{L}\right)^3}$$

$\therefore$  # electrons  $N$  is related to  $k_F, E_F$  by

$$N = 2 \frac{\frac{4}{3}\pi k_F^3}{\left(\frac{2\pi}{L}\right)^3} = \frac{k_F^3}{3\pi^2} \underbrace{L^3}_V$$

↑  
for spin

$$\frac{N}{V} = n = \frac{k_F^3}{3\pi^2}$$

$$E_F = \frac{\hbar^2}{2m} k_F^2 = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3}$$



6.

$E_F$  is a typical kinetic energy of  $e^-$  in a metal. In fact, you can show

$$\langle KE \rangle = \frac{3}{5} E_F \quad \leftarrow \text{good exercise}$$

spin  $\curvearrowright$

$$\langle KE \rangle = \frac{1}{N} 2 \int_0^{k_F} 4\pi k^2 dk \left( \frac{L}{2\pi} \right)^3 \frac{\hbar^2 k^2}{2m}$$

$$= \frac{1}{N} \left( \frac{L}{2\pi} \right)^3 8\pi \frac{\hbar^2}{2m} \frac{k_F^5}{5}$$

$$\frac{L^3}{N} \frac{1}{\pi^2} \frac{1}{5} E_F k_F^3$$

$$\frac{1}{5} E_F \frac{3\pi^2 N}{L^3}$$

$$= \frac{3}{5} E_F$$

How big is  $E_F$ ? Any guesses?

What is KE of air molecule?

$$\frac{3}{2} k_B T = \frac{3}{2} (1.38 \cdot 10^{-23} \frac{\text{J}}{\text{°K}}) (300 \text{°K})$$

$$= 6.21 \cdot 10^{-21} \text{ J}$$

7.

velocity  $\frac{1}{2}mv^2 = 6.21 \cdot 10^{-21} \text{ J}$

↑  
32 (1.67 · 10<sup>-27</sup>) kg

$$v = 4.82 \cdot 10^2 \text{ m/s}$$

Fermi Energy

$$E_F = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3}$$

$$= \frac{(1.055 \cdot 10^{-34})^2}{2(9.11 \cdot 10^{-31})} \left(3\pi^2 \frac{6}{23} 10^{29}\right)^{2/3}$$

$$= \frac{(1.055)^2}{2(9.11)} \left(\frac{3\pi^2 \cdot 600}{23}\right)^{2/3}$$

$$10^{-68} 10^{31} 10^{18}$$

$$= 5.1 \cdot 10^{-19} \text{ J}$$

$$= 3 \text{ eV}$$

$$\frac{6 \cdot 10^{23}}{23 \cdot 10^{-6}} \leftarrow \text{mole}$$

Na 23  
density 1  
cm<sup>3</sup> → m<sup>3</sup>

← MAIN POINT  
IS 100 X AS  
BIG AS KE  
OF AIR MOLECULE

Another imp't application is astrophysical...

prevents collapse of neutron star : gravity favors  
smaller radius but if  $n$  increases so does KE cost

8.

Neutron star  $N$  neutrons

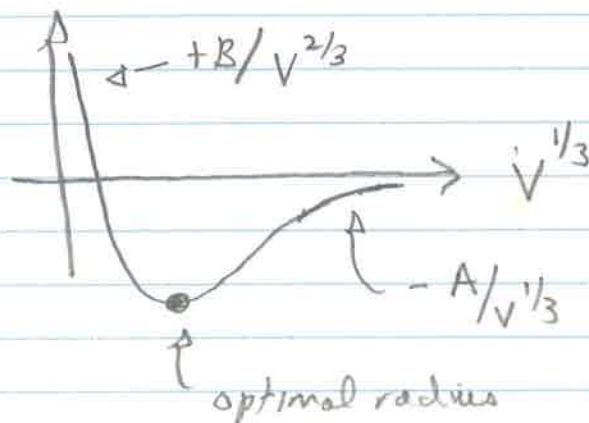
$$PE \sim -\frac{Gm^2}{r} N^2 \sim -Gm^2 N^2 V^{-1/3}$$

$\uparrow$                        $\uparrow$   
 $r$                       # pairs

typical separation  $\sim V^{1/3}$ 

$$KE \sim \frac{1}{2} N E_F = N \frac{\hbar^2}{2m} \left( 3\pi^2 \frac{N}{V} \right)^{2/3}$$

$$\sim \frac{\hbar^2}{2m} (3\pi^2)^{2/3} N^{5/3} V^{-2/3}$$

If  $V \rightarrow V/8$       PE increases by  $\times 2$ KE increases by  $\times 4$ Balance between 2  $\rightarrow$  neutron star radius.Need to know  $N$ , typical # of neutronsseen before  
where?



81

## Neutron star numbers

$$E = PE + KE$$

$$= -\frac{GM^2}{r} N^2 + \frac{\hbar^2}{2M} (3\pi^2)^{2/3} N^{5/3} \frac{1}{r^2}$$

$$G = 6.67 \cdot 10^{-11}$$

$$M = 1.67 \cdot 10^{-27}$$

$$\hbar = 1.055 \cdot 10^{-34}$$

$$0 = \frac{dE}{dr} = \frac{GM^2 N^2}{r^2} - \frac{\hbar^2 (3\pi^2)^{2/3} N^{5/3}}{M r^3}$$

$$r_0 = \frac{\hbar^2 (3\pi^2)^{2/3} N^{-1/3}}{GM^3}$$

$$\rightarrow N = \frac{2 \cdot 10^{30}}{1.67 \cdot 10^{-27}}$$

$$M_{\text{sun}} = 2 \cdot 10^{30}$$

$$r_0 = \frac{(1.055 \cdot 10^{-34})^2 (3\pi^2)^{2/3} \left(\frac{2}{1.67}\right)^{1/3} (10^{57})^{-1/3}}{6.67 \cdot 10^{-11} (1.67 \cdot 10^{-27})^3}$$

$$\sim \frac{10^{-68} 10^{-19} 10^{11} 10^{81}}{(6.67)(1.67)^3} \frac{(1.055)^2 (3\pi^2)^{2/3} \left(\frac{2}{1.67}\right)^{1/3}}{1}$$

↑

$10^5$  m

2

$1/10$

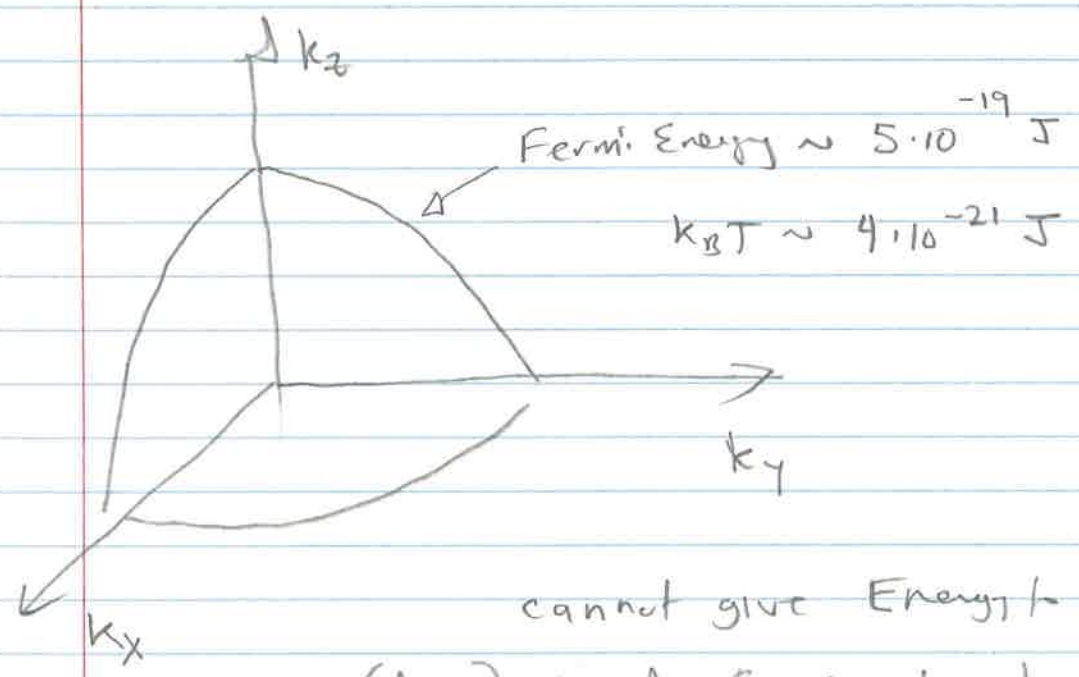
$10^4$  m  $\sim 10$  km

9.

Specific heat of solid

$$C(T) = \gamma T + AT^3$$

Crude argument



cannot give Energy to electron (deep) inside Fermi sphere because all neighboring states filled (Pauli Blocked)

only states within  $k_B T$  of  $E_F$  can absorb energy (respond to increase in T)

$$C \sim N k_B \frac{k_B T}{E_F} \quad \therefore C \sim \gamma T$$

↑ classical ideal gas answer

↑ Pauli Blocking reduction

10.

This is all at  $T \geq 0$ , assume  $e^-$  have minimum possible energy by occupying lowest states possible.

From our review of stat. mech, as  $T$  increases, we begin to occupy higher states

$E_n$   $\leftarrow$  can be occupied or empty  
but not multiply occupied!