

we found for ideal gas $H = \sum p_i^2 / 2m$ that

$$N(E) = V^N (2\pi m)^{3N/2} E^{3N/2 - 1} / \Gamma(3N/2)$$

is total number of states of energy E . A key feature of this was rapid growth with E

what do I mean. Suppose you increase energy of system of 10^{23} particles by 1 part in a trillion. How many more states are accessible

$$\begin{aligned} [1 + 10^{-12}]^{10^{23}} &= [1 + 10^{-12}]^{10^{12} \cdot 10^{11}} \\ &= e^{10^{21}} \quad \text{a lot!} \end{aligned}$$

We will come back to this again in a few minutes. First though, $N(E)$ is a less familiar concept. Can we compute something a bit more familiar?

NB However one should get familiar with $N(E)$ it is a central quantity in condensed matter physics. $\Gamma \sim \langle f | v | i \rangle^2 N(E)$

$$\left[1 + \frac{x}{n}\right]^n = e^x \quad \text{for } n \text{ large}$$

Momentum distribution function

$P(p)$ prob any particle has momentum p

$$P(p) = \langle \sum_i \delta(p_i - p) \rangle$$

Any ideas what to expect?
 $P(p)$

Canonical, get answer before doing calculation

$\langle A(x) \rangle$

$= \int p(x) A(x) dx$

// //

$\delta(E-H)$

$\int d\Gamma \delta(E-H)$

Just do $\delta(p_i - p)$

$$= \frac{1}{\int d\Gamma \delta(E-H)} \sum_i \int d\Gamma \delta(E-H) \delta(p_i - p)$$

total # of cases

cases satisfying desired condition

$$= \frac{NV^N}{\Omega(E)} \int dp_1 \dots dp_N \delta(p_1 - p) \delta[E - \frac{1}{2m}(p_1^2 + p_2^2 + \dots + p_N^2)]$$

$$= 4\pi \frac{NV^N}{\Omega(E)} p^2 \int dp_2 \dots dp_N \delta[E - \frac{p^2}{2m} - (\frac{p_2^2}{2m} + \dots + \frac{p_N^2}{2m})]$$

careful!

same integral as for $N(E)$ except $N-1$ particles and energy $E \rightarrow E - p^2/2m$

$$N(E) = \frac{V^N (2\pi m)^{\frac{3N}{2}} E^{\frac{3N}{2} - 1}}{\Gamma(\frac{3N}{2})}$$

$$= \frac{N \Gamma(\frac{3N}{2})}{\Gamma(\frac{3}{2}(N-1))} \frac{4\pi p^2}{(2\pi m)^{3/2}} \frac{1}{E^{3/2}} \left(1 - \frac{p^2}{2mE}\right)^{\frac{3}{2}(N-1)}$$

Use $\Gamma(z+1) = z \Gamma(z)$

$\lim_{n \rightarrow \infty} (1 - \frac{x}{n})^n = e^{-x}$ $E \equiv \frac{E}{N}$

$$N(p) = N \frac{4\pi p^2}{[2\pi m \frac{2}{3} E]^{3/2}} e^{-\frac{p^2}{2m} \frac{1}{2/3 E}}$$

Looks sort of familiar. We expect to find

$$N(p) = \frac{N \Gamma(3N/2)}{\Gamma(3N/2 - 3/2)} \frac{4\pi p^2}{(2\pi m)^{3/2}} \frac{1}{E^{3/2}} \left(1 - \frac{p^2}{2mE}\right)^{3/2(N-1)}$$

$$= \frac{4\pi p^2}{(2\pi m \frac{2}{3} E)^{3/2}} \frac{1}{(\frac{3}{2})^{3/2} N^{3/2}} \left(1 - \frac{p^2}{\frac{3N}{2} \frac{2mE}{3}}\right)^{3/2 N}$$

$\Gamma(z+1) = z\Gamma(z)$

$\epsilon = E/N$

$$= N \frac{4\pi p^2}{(2\pi m \frac{2}{3} \epsilon)^{3/2}} e^{-p^2/2m \cdot 1/(2/3\epsilon)}$$

← dimension / moment

$$\int_0^\infty N(p) dp = N \frac{4\pi}{(2\pi m \frac{2}{3} \epsilon)^{3/2}} \int_0^\infty p^2 e^{-p^2/2m \cdot 1/(2/3\epsilon)} dp$$

$$\int_0^\infty e^{-ax^2} dx = \frac{1}{2} \sqrt{\frac{\pi}{a}}$$

$$\int_0^\infty x e^{-ax^2} dx = \frac{1}{2} \sqrt{\frac{\pi}{a}} \cdot \frac{1}{2}$$

$$\int_0^\infty x^2 e^{-ax^2} dx = \frac{1}{2} \sqrt{\frac{\pi}{a}} \cdot \frac{3}{4}$$

$$= N \frac{4\pi}{(2\pi m \frac{2}{3} \epsilon)^{3/2}} \frac{3\sqrt{\pi}}{8} \frac{1}{(2m \frac{2}{3} \epsilon)^{3/2}}$$

$$= \frac{N}{\pi} \frac{3}{2} \frac{2m \frac{2}{3} \epsilon}{\epsilon}$$

$$= N \frac{3}{2} \frac{2m \frac{2}{3} \epsilon}{\epsilon}$$

$$N(p) \sim p^2 e^{-\frac{p^2}{2m} \frac{1}{k_B T}}$$

so we would be okay as long as $\frac{2}{3} \epsilon = k_B T$

which in fact is okay $\epsilon = \frac{3}{2} k_B T$.

Let us now consider putting 2 systems into

contact so they can exchange energy



$$H = H_1 + H_2$$

$$N(E) = \int d\Gamma \delta(H_1 + H_2 - E)$$

1 / ~~total~~ # of states with total energy E
 ← system 2 must have $E_2 = E - E_1$
 simple logic in micro-canon

Probability system 1 has energy E_1

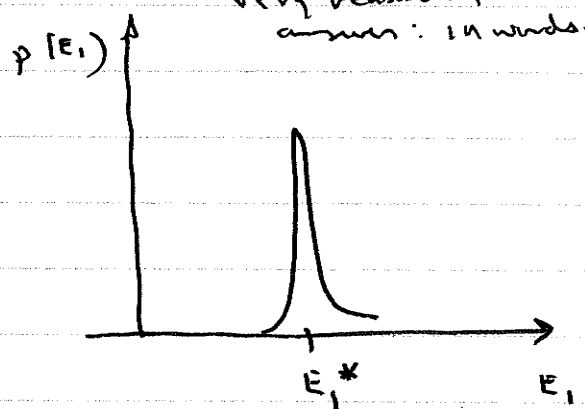
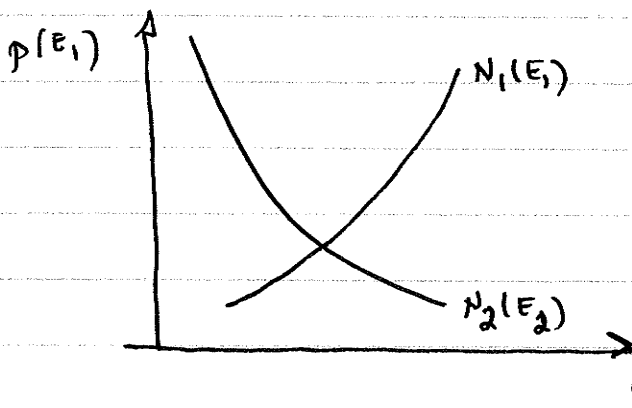
$$P(E_1) = \frac{1}{N(E)} \int d\Gamma_2 \delta(H_2 - (E - E_1))$$

$$= \frac{1}{N(E)} \int d\Gamma_1 d\Gamma_2 \delta(H_1 - E_1) \delta(H_1 + H_2 - E)$$

$$= \frac{1}{N(E)} N_1(E_1) N_2(E_2)$$

$$E_2 = E - E_1$$

Very reasonable answer: in words...



Maximize $p(E_1)$ or equivalently $\ln p(E_1)$

$$\ln p(E_1) = \ln N_1(E_1) + \ln N_2(E_2) - \ln N(E)$$

$$0 = \frac{\partial}{\partial E_1} \ln N_1(E_1) - \frac{\partial}{\partial E_2} \ln N_2(E_2)$$

$$\boxed{\frac{\partial}{\partial E_1} \ln N_1(E_1) = \frac{\partial}{\partial E_2} \ln N_2(E_2)}$$

We have discovered a quantity that must be the same for two systems in contact. we will define it to be the inverse temperature

$$\boxed{1/k_B T \equiv \frac{\partial}{\partial E} \ln N(E)}$$

How is
equilibrium
approached?

Reiterating $T_1 = T_2$ when systems in contact. Ideal gas

$$N(E) = c E^{3N/2}$$

$$\ln N(E) = \ln c + \frac{3N}{2} \ln E$$

$$1/k_B T = \frac{3N}{2E}$$

$$E = \frac{3}{2} N k_B T$$

Specific heat

$$C = \frac{dE}{dT} = \frac{3}{2} N k_B$$

Eqn of state ??? \leftarrow HW

So indeed

$$N(p) \sim p^2 e^{-p^2/2m} 1/k_B T$$

We will talk later at greater length about thermodynamics,
 but for the moment remember that $dE = TdS - pdV$

$$\text{so } \frac{1}{T} = \frac{\partial S}{\partial E} \quad \frac{1}{k_B T} = \frac{\partial}{\partial E} \ln N(E)$$

$$\rightarrow \boxed{S = k_B \ln N(E)}$$

perhaps this is familiar to you "entropy
 is the logarithm of number of states accessible
 to system"

~~May as well check out pressure~~

~~$$\frac{\partial^2 E}{\partial S^2}$$~~

~~$$T = \frac{\partial E}{\partial S} \quad p = \frac{\partial E}{\partial V}$$~~

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Before abandoning MCE for CE which is much easier to work with, let's do one more example.

N degrees of freedom S_i which can take on values ± 1

$$H = -B \sum S_i$$

$$N_{\uparrow} + N_{\downarrow} = N$$

$$E = -B(N_{\uparrow} - N_{\downarrow}) = -B(2N_{\uparrow} - N)$$

$$N(E) = \frac{N!}{N_{\uparrow}! N_{\downarrow}!}$$

$$\ln N(E) = N \ln N - N_{\uparrow} \ln N_{\uparrow} - N_{\downarrow} \ln N_{\downarrow}$$

$$\frac{1}{k_B T} = \frac{\partial}{\partial E} \ln N(E) = -\frac{\partial N_{\uparrow}}{\partial E} (1 + \ln N_{\uparrow}) - \frac{\partial N_{\downarrow}}{\partial E} (1 + \ln N_{\downarrow})$$

We also know that

$$\frac{\partial N_{\downarrow}}{\partial E} = -\frac{\partial N_{\uparrow}}{\partial E} \quad \text{since } N_{\uparrow} + N_{\downarrow} = N \text{ a constant}$$

$$\frac{\partial N_{\uparrow}}{\partial E} = -\frac{1}{2B}$$

$$\frac{1}{k_B T} = +\frac{1}{2B} (1 + \ln N_{\uparrow}) - \frac{1}{2B} (1 + \ln N_{\downarrow})$$

$$\ln(N_{\uparrow}/N_{\downarrow}) = e^{2B/k_B T}$$

$$N_{\uparrow}/N_{\downarrow} = e^{2B/k_B T}$$

This should again be familiar since we have
 two possible choices of Energy $-B$ for $S_i = \uparrow = 1$
 and $+B$ for $S_i = \downarrow = -1$ and the difference is
 then $2B$ so we learn relative #s of 2 possibilities

$$\sim e^{-\Delta E/k_B T}$$

Summarize ME

1) $P(\Gamma) = \text{same for all } \Gamma \text{ of energy } E$

Formally $P(\Gamma) = \frac{\delta(E-H)}{\int d\Gamma \delta(E-H)}$
 \uparrow just check if $H(\Gamma) = E$ \uparrow divide by total # of Γ satisfying this

2) $N(E) = \int d\Gamma \delta(E-H)$

3) $1/k_B T = \partial/\partial E \ln N(E)$ $T_1 = T_2$

4) $S = k_B \ln N(E)$

Ideal gas example, spin $1/2$ example

Now relate to canonical ensemble

Imagine system 1 is in contact with some much larger system 2. Consider computing the average of some quantity depending on variables in 1.

$$\langle \hat{O}_1 \rangle = \frac{1}{N(E)} \int d\Gamma_1, d\Gamma_2 \hat{O}_1 \delta(H_1 + H_2 - E) \dots$$

$$N(E) = \int d\Gamma_1, d\Gamma_2 \delta(H_1 + H_2 - E)$$

$$\langle \hat{O}_1 \rangle = \int d\Gamma_1 \hat{O}_1 \frac{\int d\Gamma_2 \delta(H_1 + H_2 - E)}{\int d\Gamma_1, d\Gamma_2 \delta(H_1 + H_2 - E)}$$



$P_1(\Gamma_1) \equiv$ factor associated with "integrating out" degrees of freedom of system 2

This factor $\sim N_2(E - H_1)$

$$\ln N_2(E - H_1) = \ln N_2(E) - H_1 \frac{\partial}{\partial E} \ln N_2(E) + \frac{1}{2} H_1^2 \frac{\partial^2}{\partial E^2} \ln N_2(E) \dots$$

since $H_1 \ll E$ is small

↑
 $\frac{1}{k_B T} \quad T_2$

Compare terms: Ideal gas $N_2(E) \sim E^{3N/2}$

$$\frac{\partial}{\partial E} \ln N_2(E) = \frac{3N}{2E}$$

$$\frac{\partial^2}{\partial E^2} \ln N_2(E) \sim -\frac{3N}{2E^2}$$

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$$\text{so } \ln N_2(E) = o\left(\frac{H_1}{E}\right) + o\left(\frac{H_1^2}{E^2}\right)$$

$$\ln N_2(E - H_1) = C - H_1/k_B T$$

$$N_2(E - H_1) = \tilde{c} e^{-H_1/k_B T} \quad \therefore p(\Gamma_1) \propto e^{-H_1/k_B T}$$

$$\langle \hat{O}_1 \rangle = \frac{\int d\Gamma_1 \hat{O}_1 e^{-H_1/k_B T}}{\int d\Gamma_1 e^{-H_1/k_B T}}$$

↙ must be this to get
normalization correct
 $Z(\beta) \quad \beta = 1/k_B T$

Aside: Shannon argument.

"If you know nothing about probability distribution, make all probabilities equal subject to $\sum p_i = 1$ "

["Maximum Entropy" method in numerical simulations]

"If you know a little make broadest prob. distrib. consistent with that knowledge"

[Again Maxent - know the moments \bar{A} and know $G(p, \tau)$ from Monte Carlo at some points]

Shannon's claim is maximize

$$\sum p_i \ln p_i \quad \text{subject to} \quad \sum p_i = 1 \quad \sum p_i A_i = \langle A \rangle$$

accomplishes this

$$f = \sum p_i \ln p_i + \lambda_1 (\sum p_i - 1) + \lambda_2 (\sum p_i A_i - \langle A \rangle)$$

$$\frac{\partial}{\partial p_i} f = \ln p_i + 1 + \lambda_1 + \lambda_2 A_i = 0$$

$$p_i = e^{-\lambda_1 - 1} e^{-\lambda_2 A_i}$$

$$\sum p_i = 1 \quad \rightarrow \quad p_i = \frac{e^{-\lambda_2 A_i}}{\sum e^{-\lambda_2 A_i}}$$

This is basically what our deriv of CE told us.