

Electrons in solids

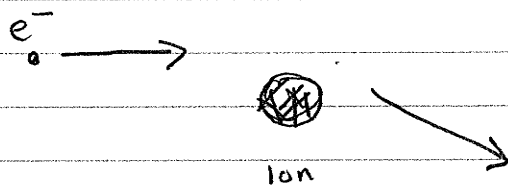
We have focussed so far on lattice

- lattice structure
- description of x ray scattering
- lattice vibrations

Turn now to electrons

An amazing thing is that a lot of properties of e^- in solid can be explained by ignoring lattice!

How is it possible electrons don't scatter off ions?!



Recall QM

$$\psi(\vec{r}, 0) \longrightarrow \psi(\vec{r}, t) \quad \text{How?}$$

Solve

$$\left[-\frac{\hbar^2}{2m} \nabla^2 + V(r) \right] \phi_n(\vec{r}) = E_n \phi_n(\vec{r})$$

↑ Hermitian

↑ complete set
↓ orthogonal

Expand

$$\psi(\vec{r}, 0) = \sum_n c_n \phi_n(\vec{r})$$

$$c_n = \int d^3\vec{r} \psi(\vec{r}, 0) \phi_n^*(\vec{r})$$

2.

$$\psi(\vec{r}, t) = \sum_n c_n \phi_n(\vec{r}) e^{-iE_n t/\hbar}$$

Sometimes eliminate c_n and write in fancy way

$$\psi(\vec{r}, t) = \int d^3 r' \left[\sum_n \phi_n^*(r') \phi_n(r) e^{-iE_n t/\hbar} \right] \psi(\vec{r}', 0)$$

$G(\vec{r}, \vec{r}', t)$ "Green's function"
or "propagator"

But point to emphasize is that if you have e^-

in one of eigenstates initially $\psi(\vec{r}, 0) \equiv \phi_e(\vec{r})$

if never scatters out of it $\psi(\vec{r}, t) = \phi_e(r) e^{-iE_e t/\hbar}$

[Analogous in a way to classical normal modes]

One has found special states (by "absorbing"

effect of ions into appropriate $\phi_n(\vec{r})$) ~~where~~

"LANDAU FERMI LIQUID THEORY"

e^- in solid can be understood as free electrons
(with perhaps renormalized $m \rightarrow m_*$)

Perhaps
most famous
principle of
CM physics

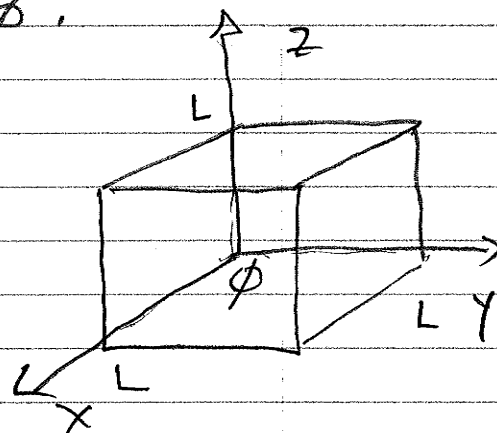
3,

We will see other ways to understand this better in the coming weeks.

For now let's see what comes out of a description of e^- in a solid as just particles in a box with no ionic potential $V(\vec{r}) = \phi$.

$$-\frac{\hbar^2}{2m} \nabla^2 \phi_n = E_n \phi_n$$

$$\frac{d^2}{dx^2} + \frac{d^2}{dy^2} + \frac{d^2}{dz^2}$$



Solns are

$$\phi_n(x, y, z) = \sin \frac{n_x \pi x}{L} \sin \frac{n_y \pi y}{L} \sin \frac{n_z \pi z}{L}$$

$n_x \ n_y \ n_z$

Can easily check this obeys Sch Eqn with

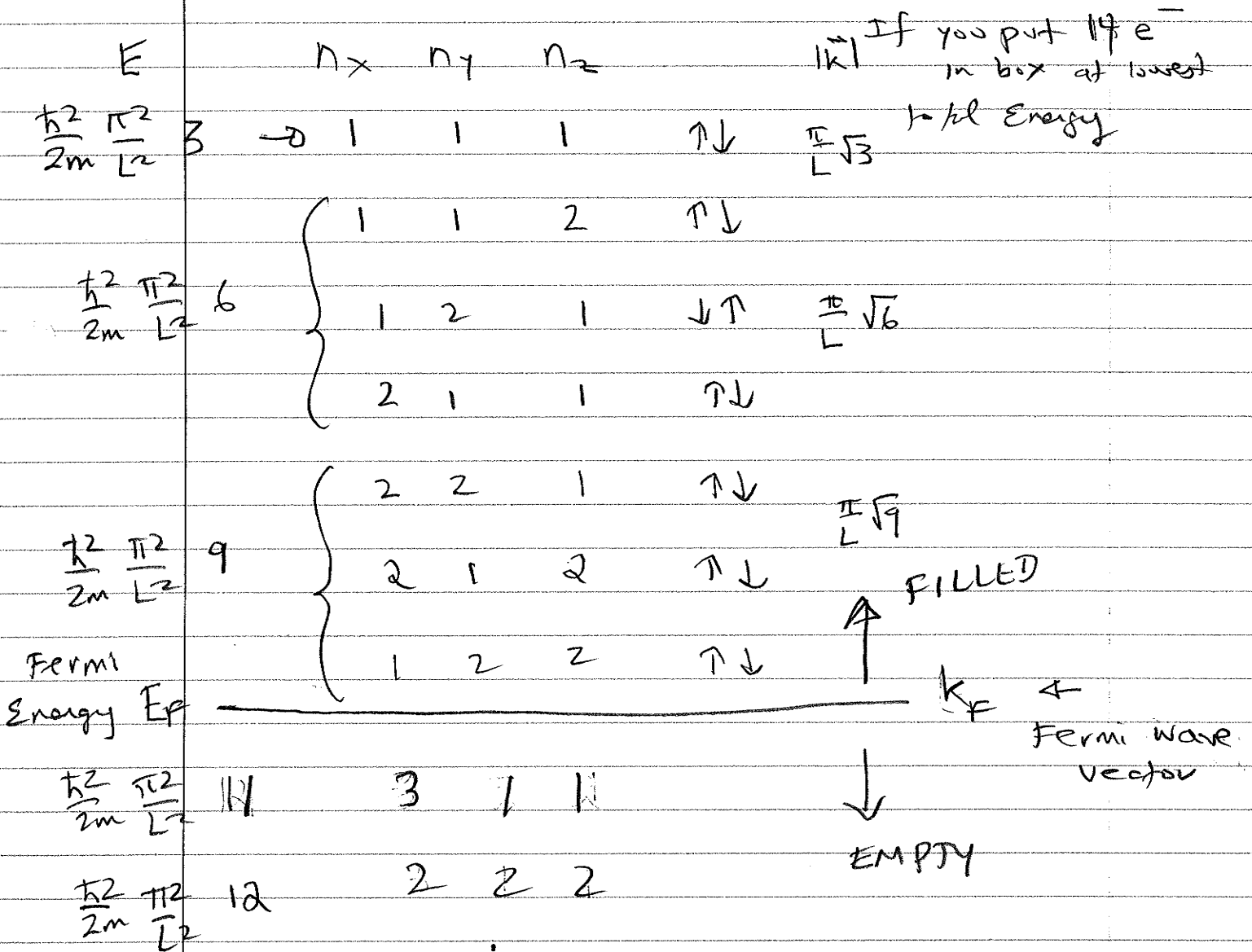
$$E_n = \frac{\hbar^2}{2m} \frac{\pi^2}{L^2} (n_x^2 + n_y^2 + n_z^2)$$

$$k_x = \frac{\pi n_x}{L} \quad k_y = \frac{\pi n_y}{L} \quad k_z = \frac{\pi n_z}{L}$$

As $L \rightarrow \infty$ k_x, k_y, k_z continuous.

4.

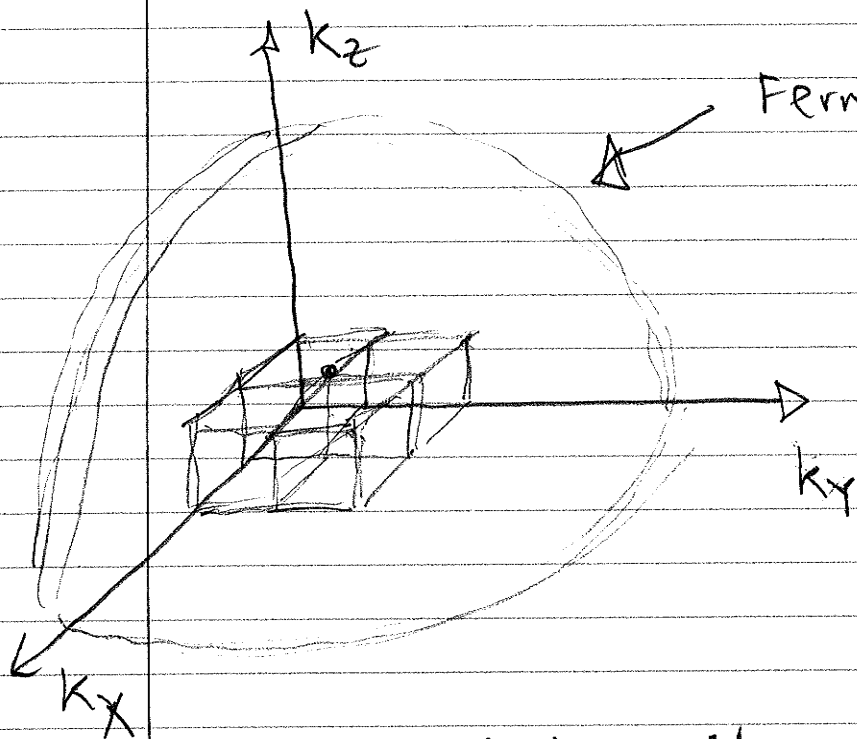
Pauli Principle: For Fermions each QM state can be occupied by at most 1 particle. (or 2 if one includes spin $\uparrow \downarrow$ as additional quantum number like $n_x n_y n_z$)



$$E_F = \frac{\hbar^2}{2m} k_F^2$$

5.

There is a \vec{k} point in every $\left(\frac{2\pi}{L}\right)^3$ of \vec{k} space



points inside

$$= \frac{\frac{4}{3} \pi k_F^3}{\left(\frac{2\pi}{L}\right)^3}$$

\therefore # electrons N is related to k_F, E_F by

$$N = 2 \frac{\frac{4}{3} \pi k_F^3}{\left(\frac{2\pi}{L}\right)^3} = \frac{k_F^3}{3\pi^2} \underbrace{L^3}_V$$

↑
for spin

$$\frac{N}{V} = n = \frac{k_F^3}{3\pi^2}$$

$$E_F = \frac{\hbar^2}{2m} k_F^2 = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3}$$

6.

E_F is a typical kinetic energy of e^- in a metal. In fact, you can show

$$\langle KE \rangle = \frac{3}{5} E_F \quad \leftarrow \text{good exercise}$$

spin \curvearrowright

$$\langle KE \rangle = \frac{1}{N} 2 \int_0^{k_F} 4\pi k^2 dk \left(\frac{\hbar}{2\pi} \right)^2 \frac{k^2}{2m}$$

$$= \frac{1}{N} \left(\frac{\hbar}{2\pi} \right)^3 8\pi \frac{\hbar^2}{2m} \frac{k_F^5}{5}$$

$$\frac{\hbar^3}{N} \frac{1}{\pi^2} \frac{1}{5} E_F k_F^3$$

$$\frac{1}{5} E_F \frac{3\pi^2 N}{L^3}$$

$$= \frac{3}{5} E_F$$

How big is E_F ? Any guesses?

What is KE of air molecule?

$$\frac{3}{2} k_B T = \frac{3}{2} (1.38 \cdot 10^{-23} \frac{J}{K}) (300 K)$$

$$= 6.21 \cdot 10^{-21} J$$

7.

Velocity $\frac{1}{2}mv^2 = 6.21 \cdot 10^{-21} \text{ J}$

↑
 $32(1.67 \cdot 10^{-27}) \text{ kg}$

$$v = 4.82 \cdot 10^2 \text{ m/s}$$

Fermi Energy

$$E_F = \frac{\hbar^2}{2m} (3\pi^2 n)^{2/3}$$

$$= \frac{(1.055 \cdot 10^{-34})^2}{2(9.11 \cdot 10^{-31})} \left(3\pi^2 \frac{6}{23} 10^{29}\right)^{2/3}$$

$$= \frac{(1.055)^2}{2(9.11)} \left(\frac{3\pi^2 \cdot 600}{23}\right)^{2/3}$$

$$10^{-68} 10^{31} 10^{18}$$

$$= 5.1 \cdot 10^{-19} \text{ J}$$

$$= 3 \text{ eV}$$

$$\frac{6 \cdot 10^{23}}{23 \cdot 10^{-6}} \leftarrow \text{mole}$$

Na 23
density 1
cm³ → m³

← MAIN POINT
 IS 100x AS
 BIG AS KE
 OF AIR MOLECULE

Another imppt application is astrophysical...

prevents collapse of neutron star : gravity favors
 smaller radius but if n increases so does KE cost

8.

Neutron star N neutrons

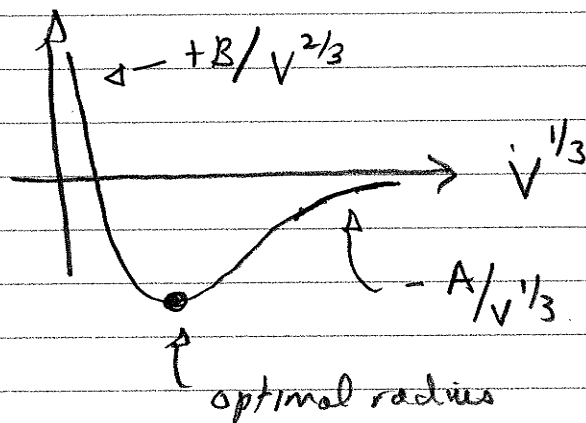
$$PE \sim - \frac{Gm^2}{r} N^2 \sim - Gm^2 N^2 V^{-1/3}$$

\uparrow \uparrow
 r # pairs

typical separation $\sim V^{1/3}$

$$KE \sim \frac{1}{2} N E_F = N \frac{\hbar^2}{2m} \left(3\pi^2 \frac{N}{V} \right)^{2/3}$$

$$\sim \frac{\hbar^2}{2m} (3\pi^2)^{2/3} N^{5/3} V^{-2/3}$$

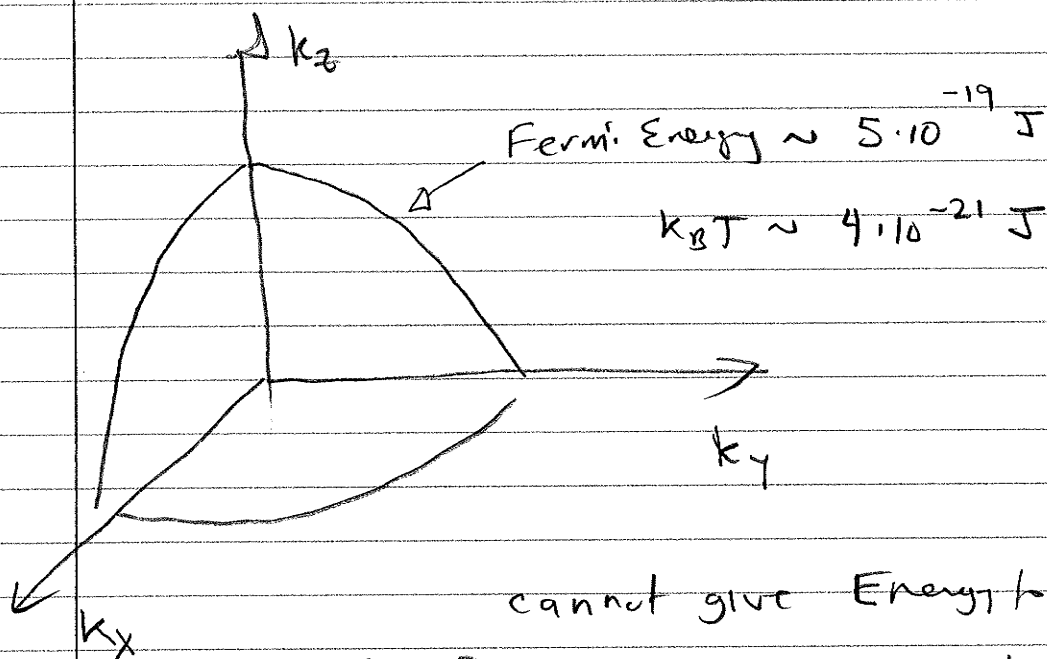
If $V \rightarrow V/8$ PE increases by $\times 2$ KE increases by $\times 4$ Balance between 2 \rightarrow neutron star radiusNeed to know N , typical # of neutronsseen before
where?

9.

Specific heat of solid

$$C(T) = \gamma T + AT^3$$

Crude argument



cannot give Energy to electron
(deep) inside Fermi sphere because all
neighboring states filled (Pauli Blocked)

Only states within $k_B T$ of E_F can
absorb energy (respond to increase in T)

$$C \sim N k_B \frac{k_B T}{E_F} \quad \therefore C \sim \gamma T$$

↑
classical
ideal gas
answer

↑
Pauli
Blocking reduction